

## Retrosynthetic Analysis of Fullerene C<sub>60</sub>: Structure, Stereochemistry, and Calculated Stability of C<sub>30</sub> Fragments

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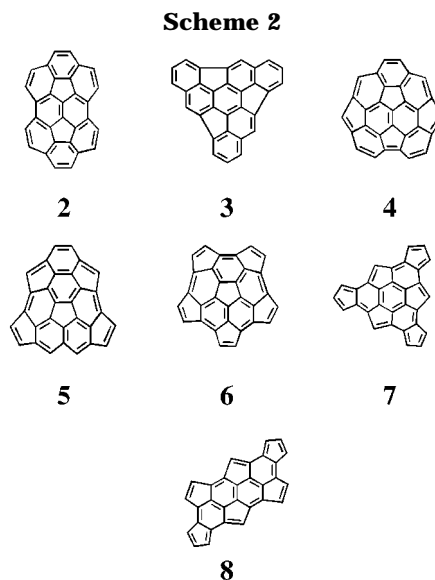
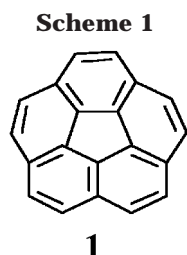
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**Abstract:** The total number of possible retrosynthetic bisections of C<sub>60</sub> leads to nine different isometric C<sub>30</sub> fragments. These molecules include five chiral units, four of which derive from partitions corresponding to four distinct "Coupes du Roi". The energies, curvatures, and homodesmotic stabilization energies of the C<sub>30</sub> fragments are evaluated at the ab initio 6-31G level.

Bowl-shaped polyaromatic hydrocarbons, which are fragments of the fullerene structure (*buckybowls*, after the American architect R. Buckminster Fuller), have generated intense research activity in recent years.<sup>1–3</sup> Most of such curved hydrocarbons have been synthesized in the recent years, although the recently improved<sup>4</sup> syntheses of the smallest of these curved fragments, corannulene **1** (Scheme 1), was first described long before<sup>5</sup> the discovery of fullerenes.

The interest in these curved buckybowls, involving two distinct (concave and convex) surfaces, includes various aspects related to properties such as, e.g., reactivity<sup>1,6</sup> or endo- vs exohedral metal complexation ability,<sup>7</sup> as well as other theoretical aspects of these interesting conjugated and strained structures.<sup>8</sup> It is also interesting to



note that curvature induces chirality for some of these bowl-shaped hydrocarbons.

As far as fullerene C<sub>60</sub> is considered, the use of these buckybowls as synthons for some future total syntheses might also be imagined. In fact, several such C<sub>30</sub> fragments (**2–8**) have been recognized<sup>2,3,9–11</sup> (Scheme 2), and some have been prepared: **2**<sup>12</sup> and **3**.<sup>13,14</sup>

These molecules include (i) a nonisometric<sup>15</sup> fragment **2**; (ii) a chiral buckybowl **8** (generated by a homochiral<sup>16</sup> segmentation usually referred to as "La Coupe du Roi");<sup>17,18</sup> (iii) a chiral isomer **3** (coming from a heterochiral<sup>16</sup> bisection); and (iv) four achiral isomers (**4–7**).

However, the total number of all possible isometric C<sub>30</sub>

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(15) Two molecules are isometric if the atoms and their connectivity are the same. Thus, the *nonisometric* buckybowl **2** (Scheme 2) "...is not an exact half of C<sub>60</sub> since when buckminsterfullerene is 'taken apart' to produce the carbon framework that leads to it, the remaining carbon fragment is not identical with it....", as stated in ref 2.

(16) A brief and interesting glossary of stereochemical terminology is found in the ref 17.

(17) For a nice historical account of the trick to divide an apple into two homochiral halves by "La Coupe du Roi" and for the use of this story in chemistry, see: Mislow, K. *Bull. Soc. Chim. Fr.* **1994**, *131*, 534.

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fragments, either chiral or achiral, has not yet been methodically identified.

The general problem of fragmenting molecules into isometric segments, including the bisection of an achiral object into isometric homochiral halves (La Coupe du Roi: a way to divide an apple into two halves that have the *same* chirality),<sup>16</sup> has been comprehensively examined by Mislow.<sup>17</sup> Therefore, we decided to use this methodology<sup>17,18</sup> to determine the total number of pairs of isometric C<sub>30</sub> fragments constituting the fullerene C<sub>60</sub>.

We find *nine* bisections, including four chiral C<sub>30</sub> buckybawls from four different Coupes du Roi. Additional insight into this series is also provided by (i) the calculated<sup>19</sup> energies and curvatures of these C<sub>60</sub> fragments and (ii) the estimated strain energies, reflected by the stabilization energies obtained from the appropriate homodesmotic reactions.

Mislow and co-workers<sup>17</sup> pointed out several limitations imposed on the ways in which molecules can be segmented. First, some of their assumptions will be briefly recalled before the segmentation of C<sub>60</sub> into isometric halves is considered. Thus, the cut objects (C/O) are defined as "the ensemble of isometric segments in the shape of the object with the cut in place". The segments must be related by a symmetry operation of the C/O. Consequently, as long as achiral molecules are considered, bisections may yield achiral, heterochiral, and homochiral segments.

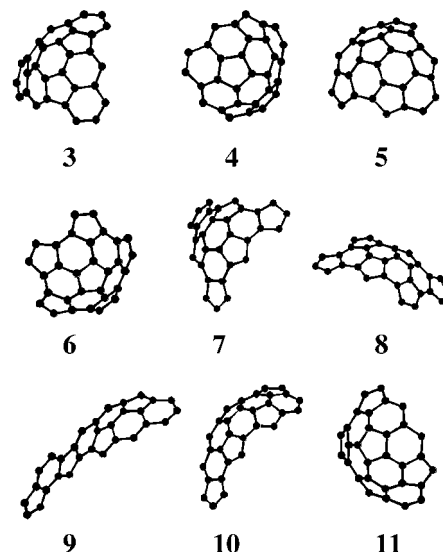
Considering the icosahedral object C<sub>60</sub> (symmetry elements corresponding to a truncated regular icosahedron) and the dissections<sup>20</sup> to C<sub>30</sub> segments, the connections of *two isometric halves*, which have to be related by a symmetry operation of the C/O, *involve exclusively a C<sub>2</sub> axis or an inversion center*. The corresponding "belts"<sup>21</sup> can also be identified by applying the same two symmetry elements.

Practically, belts are obtained starting at *any pair of condensed rings* on the surface of C<sub>60</sub> and, with the consideration of these two symmetry elements, by achieving one turn around the fullerene.<sup>22</sup> The distinct dissections, and consequently the C<sub>30</sub> segments, eventually arise from these belt structures.

Excluding the nonisometric isomer **2**, we presently find nine bisections to isometric C<sub>30</sub>, that is, three additional bisections as compared to the previous ones (Scheme 2). The present nine C<sub>30</sub>, including heterochiral fragment **3**, as well as four homochiral (**8–11**) and four achiral segments (**4–7**), are shown in the Figure 1.

We have calculated at the ab initio RHF 6-31G level<sup>23</sup> the relative energies of energy-minimized geometries of these nine C<sub>30</sub> segments,<sup>24</sup> as well as the parameters related to their curvatures.<sup>22</sup> The results involve three series of compounds (Table 1).

Interesting tendencies are displayed by these series: (i) the more stable isomers (**3** and **4**) correspond to the



**Figure 1.** Nine isometric C<sub>30</sub> energy-minimized structures. Fragments **3** (resulting from a heterochiral cut) and **8–11** are chiral.

lower number of five-membered rings; (ii) within each (C<sub>30</sub>H<sub>10</sub>) and (C<sub>30</sub>H<sub>12</sub>), series the energy contents smoothly increase when the number of pentagons is increased by one, although a more abrupt variation occurs if the number of pentagons is simultaneously increased by two units (compare **4** to **5** vs **3** to **10**); and (iii) the shapes of the segments differ considerably from the carbon skeletons mapped onto the C<sub>60</sub> surface prior to segmentation. This issue, involving the "curvatures" of these molecules, is now detailed.

The pyramidalization angle obtained from the  $\pi$ -orbital axis vector (POAV) has been shown to provide a useful index of the "local curvature" of fullerenes.<sup>25</sup> Alternatively, the present calculation of dihedral angles  $\theta$  (as defined in the caption of Figure 2;  $\theta_a$  and  $\theta_{max}$  are, respectively, average and maximum values) also yields useful local parameters to characterize the curvature of such nonplanar polyaromatic hydrocarbons. In fact, for every distinct C<sub>30</sub>, there are significant variations of curvature along the structure,<sup>22</sup> emphasized by the ( $\theta_{max}$  –  $\theta_a$ ) differences (Table 1). For comparison, the "regularly curved" spherical fullerene C<sub>60</sub> corresponds to  $\theta_{max} = \theta_a = 41.8^\circ$ . These "irregular" distributions of curvature are reminiscent of the recent gas-phase molecular structure of corannulene, exhibiting a slightly flattened shape.<sup>26</sup> In addition, the curvature values of the distinct C<sub>30</sub>

(23) Geometries were optimized and energies computed at the ab initio RHF 6-31G level using the Gaussian 98 (revision A.7) package: Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Zakrzewski, V. G.; Montgomery, J. A., Jr.; Stratmann, R. E.; Burant, J. C.; Dapprich, S.; Millam, J. M.; Daniels, A. D.; Kudin, K. N.; Strain, M. C.; Farkas, O.; Tomasi, J.; Barone, V.; Cossi, M.; Cammi, R.; Mennucci, B.; Pomelli, C.; Adamo, C.; Clifford, S.; Ochterski, J.; Petersson, G. A.; Ayala, P. Y.; Cui, Q.; Morokuma, K.; Malick, D. K.; Rabuck, A. D.; Raghavachari, K.; Foresman, J. B.; Cioslowski, J.; Ortiz, J. V.; Stefanov, B. B.; Liu, G.; Liashenko, A.; Piskorz, P.; Komaromi, I.; Gomperts, R.; Martin, R. L.; Fox, D. J.; Keith, T.; Al-Laham, M. A.; Peng, C. Y.; Nanayakkara, A.; Gonzalez, C.; Challacombe, M.; Gill, P. M. W.; Johnson, B. G.; Chen, W.; Wong, M. W.; Andres, J. L.; Head-Gordon, M.; Replogle, E. S.; Pople, J. A. *Gaussian 98*, revision A.7; Gaussian, Inc.: Pittsburgh, PA, 1998.

(24) The relative energies of these fragments have also been computed at the semiempirical AM1 level, and very similar trends to the ab initio results are obtained.<sup>22</sup>

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(19) See also: Alvarez, S.; Serratos, F. *J. Am. Chem. Soc.* **1992**, *114*, 2623

(20) The "cuts" corresponding to various bisections involve homolytic bond splitting. The C<sub>30</sub> segments are obtained by attachment of hydrogen atoms to the resulting "dangling" carbon bonds.

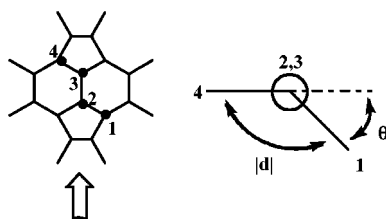
(21) Sequences of five- and/or six-membered condensed rings on the C<sub>60</sub> surface outline the "belts",<sup>22</sup> assembled from the end-carbons of the C<sub>30</sub> segments. To outline exclusively C<sub>30</sub> fragments originating from condensed rings, the corresponding belt should only involve two-ring junctions. The "cuts", defining the split bonds of C<sub>60</sub>, cross through the junctions of the "belts".

(22) See Supporting Information.

**Table 1.** Ab initio Calculated Energies ( $\Delta E$ )<sup>a</sup> and Curvatures<sup>b</sup>  $\theta_a$  (average) and  $\theta_{\max}$  (maximum value) of Distinct Isometrical C<sub>30</sub> Segments

C <sub>30</sub>	$\Delta E$	$\theta_{\max}$	$\theta_a$	$\theta_{\max} - \theta_a$	total number of five-membered rings	total number of six-membered rings <sup>c</sup>
(C <sub>30</sub> H <sub>10</sub> )						
<b>4</b>	0	45.5	34.5	11.0	4	7
↓	4.67	45.5	33.1	12.4	5	6
<b>5</b>	5.91	43.5	33.3	10.2	5	6
↓	7.22	39.6	31.5	8.1	6	5
<b>11</b>						
↓						
<b>6</b>						
(C <sub>30</sub> H <sub>12</sub> )						
<b>3</b>	0	38.1	27.0	11.1	3	7
↓	87.90	39.0	19.6	19.4	5	5
<b>10</b>	89.18	39.4	18.2	21.2	6	4
↓	106.78	36.7	19.0	17.7	6	4
<b>8</b>						
↓						
<b>7</b>						
(C <sub>30</sub> H <sub>14</sub> )						
<b>9</b>		25.3	10.2	15.1	4	5
fullerene						
C <sub>60</sub>		41.8	41.8	0	12	20

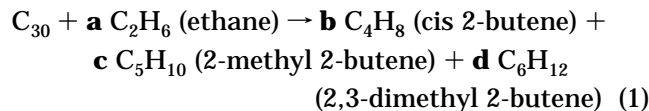
<sup>a</sup> Zero energies (in kcal/mol) correspond, in each series of compounds, to the most stable energy-minimized C<sub>30</sub> ( $\Delta E$  in kcal/mol); for the individual energy contents see ref 22. <sup>b</sup> Angles  $\theta$  (in degrees) as parameters for the C<sub>30</sub> curvatures are defined in the Figure 2 caption (see also ref 22 for all calculated values). <sup>c</sup> This number of six-membered rings does not reflect the number of benzenoid hexagons (Fries number), vide infra.



**Figure 2.** “Curvature” angles  $\theta$  are determined as  $\theta = \pi - |d|$ , with the dihedral angles  $d$  being  $\angle(1,2,3);(2,3,4)$ .

within each series are also highly dissimilar<sup>22</sup> [see the unlike mean values ( $\theta_a$ ) and disparate ( $\theta_{\max}$ ) - ( $\theta_a$ ) differences (Table 1)]. Moreover, the shapes of the C<sub>30</sub> are, on the average, much more flattened compared to the substructures on the C<sub>60</sub> surface before bisections (significantly lower values of  $\theta_a$  as compared to 41.8° for C<sub>60</sub>). This effect possibly reflects some release of strain upon fragmentation. However, the main effect remains the number of five-membered rings, as shown by the more stable although less flattened (higher  $\theta_a$ ) fragments (**3** and **4**) containing the lower number of pentagons.

The relative stabilization energies (SE) of this series of C<sub>30</sub> hydrocarbons at the ab initio RHF 6-31G level are compared by means of homodesmotic bond separation reactions of the type illustrated in eq 1.



As a delicate balance between  $\pi$ -conjugation stabilization and curvature-related strain destabilization, the calculated SE values (Table 2) involve surprisingly a limited range of 10 kcal/mol (around 65 kcal/mol) for most of the compounds. Two remarkable exceptions (the highly stabilized compound **3** and the more destabilized compound **4**) must be emphasized. More qualitatively, it is interesting to mention that the number of benzenoid hexagons, the so-called Fries number, which has been proposed as a  $\pi$ -system's relative-stability criterion,<sup>27</sup> is

**Table 2.** Stabilization Energies for Distinct Isometrical C<sub>30</sub> Segments<sup>a</sup>

	C <sub>30</sub> + <b>a</b> C <sub>2</sub> H <sub>6</sub> → <b>b</b> C <sub>4</sub> H <sub>8</sub> + <b>c</b> C <sub>5</sub> H <sub>10</sub> + <b>d</b> C <sub>6</sub> H <sub>12</sub>				SE
	(a)	(b)	(c)	(d)	
<b>3</b>	24	3	6	6	-144.58
<b>6</b>	25	5	0	10	-70.15
<b>4</b>	25	3	4	8	-69.23
<b>9</b>	23	2	10	3	-65.86
<b>5</b>	25	3	4	8	-64.57
<b>11</b>	25	3	4	8	-63.32
<b>8</b>	24	4	4	7	-59.47
<b>10</b>	24	2	8	5	-52.60
<b>7</b>	24	3	6	6	-37.80

<sup>a</sup> Stabilization energy SE =  $E_{\text{C}_{30}} + \mathbf{a} E_{\text{C}_2\text{H}_6} - \mathbf{b} E_{\text{C}_4\text{H}_8} - \mathbf{c} E_{\text{C}_5\text{H}_{10}} - \mathbf{d} E_{\text{C}_6\text{H}_{12}}$  in kcal/mol, using homodesmotic reactions of the type illustrated in eq 1. For the individual energy contents calculated at the ab initio RHF 6-31G level see ref 22.

also significantly higher for **3** than **7** (seven and one, respectively). It is also worth noting that the more stable fragment **3** is presently the only member of this series that has yet been synthesized.<sup>13,14</sup>

In summary, we have identified all of the possible ways to bisect fullerene C<sub>60</sub> into isometric C<sub>30</sub> segments, most of which (five out of nine) are chiral molecules. The shapes of most of these C<sub>60</sub> substructures are found to be far from perfect “geodesic” half spheres, and the associated energy values are highly dependent on the considered structures. The evaluation of the relative stabilization energies highlights the more stable, already isolated C<sub>30</sub> **3**. However, besides their strain contents, the stereochemistry of these synthons will ultimately have to be considered for future “dimerizations” in the direction of a total synthesis of C<sub>60</sub>.

**Supporting Information Available:** Dihedral angles, Cartesian coordinates, ab initio and AM1 calculated electronic energy contents, and the structures of belts corresponding to the nine distinct C<sub>30</sub> fragments. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(27) See, for example: Austin, S. J.; Fowler, P. W.; Manopoulos, D. E.; Zheng, M. *Chem. Phys. Lett.* **1994**, *228*, 478 and references therein.